1,024,869

DRAWINGS ATTACHED.

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1,024,869



Date of filing Complete Specification (under Section 3(3) of the Patents Act, 1949): Sept. 7, 1964.

Application Date: Oct. 1, 1963. No. 38522/63.

Application Date: March 20, 1964. No. 11836/64.

Complete Specification Published: April 6, 1966.

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Index at Acceptance:—G1 N(1A3B, 1C, 1D13, 1F, 1M, 3S 3, 3S7A1). Int. Cl.:—G 08 c.

COMPLETE SPECIFICATION.

Improvements in or relating to Apparatus for the Determination of the Thermal Conductivity of Gases.

We, THE DISTILLERS COMPANY LIMITED, of 12 Torphichen Street, Edinburgh 3, Scotland, a British Company, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

The present invention relates to apparatus for use in the determination of the thermal conductivity of gases.

As is well known the measurement of the thermal conductivity of gases provides one of the most widely used physical methods of gas detection. The most usual arrangement employed comprises flowing the gas stream over a heated wire whose resistance varies with temperature, which in turn depends on the thermal conductivity of the gas. The wire 20 is formed as one arm of a Wheatstone Bridge and the change in its resistance due to the cooling effect of the gas is observed as the degree of unbalance of the bridge. In a further embodiment of this method the gas stream may be caused to pass over two heated wires forming opposite arms of a Wheatstone Bridge, while a reference gas is passed over two wires forming the other two arms of the bridge. This particular method 30 of gas analysis has now assumed a new importance in the field of gas chromatography and, particularly in association with gas chromatography apparatus employing capillary columns, has necessitated new designs of thermal conductivity detectors of increased sensitivity and speed of response.

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In thermal conductivity detectors employed hitherto the cells have usually been holes drilled in a metal block and containing the heated wire filaments. Where such a detector is employed to analyse the gases issuing from a chromatography column it has been found that the smaller the volume of the cell the better is the discrimination of the detector relative to each individual gas leaving the column. With chromatography apparatus using capillary columns, detector cells having a volume of only a few microlitres are most desirable, and it is the manufacture of cells of such small volume which has hitherto proved impossible.

It has now been found possible by application of a new method of contruction to produce a cell of such a small volume. At normal capillary column flowrates, i.e. of the order of 1ml. per minute, the cell thus constructed is similar in sensitivity to the flame ionisation detector which involves a much more elaborate electrical circuit. Further advantages over flame ionisation are that the apparatus of the present invention is readily made flame-proof and is not destructive of samples.

Accordingly the present invention is apparatus for use in the determination of the thermal conductivity of gases which comprises two blocks of a thermally conductive material separated by a perforated gasket, the assembly of blocks and gasket defining one or more enclosed cells having inlet and outlet ports, with substantially the total gas space of the cell lying within the thickness of the gasket, and at least one

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electrically heated detector element in each

The blocks may be fabricated of any suitable thermally conductive material e.g. metal and in particular are fabricated of stainless steel or Duralumin.* Where the gases to be analysed are of a corrosive nature the faces of the blocks forming the sides of the cell may be plated with a corrosion resistant 10 meial e.g. platinum, rhodium, silver etc.

The gasket may also be fabricated of any suitable material which when held firmly between the blocks provides a gas tight seal.

A suitable material is for example aluminium. The gasket may be formed as a laminar structure consisting of the main body sandwiched between two foils. The gasket is perforated in any suitable manner and usually has from one to four holes of any suitable configuration e.g. slots which, when the gasket is placed between the blocks, define with the faces of the blocks the conductivity cell. The dimensions of the perforations in the gasket may be so chosen as to provide cells of any volume down to the order of 1 microlitre.

The assembly of blocks and gasket may be held together for example by clamps or by screws or bolts passing through the assembly. The clamping together of the blocks must be carefully and evenly carried out, and this, combined with flat mating surfaces and accurate construction of the parts, produces a cell which is absolutely gas tight without any sealing material. However, the sealing is facilitated by the use of the

subsidiary foil gaskets.

The detector element may be formed as a wire of for example platinum or other wire with a suitably high temperature co-efficient of resistance, of thickness down to the order of 0.0001 of a inch. This may be supported by two wires of noble metal alloy which pass through insulators in one of the blocks and project from the block to a distance equal to half the thickness of the gasket system. The detector element is soldered or welded to the ends of these wires. One of these wires is free for some of its length within the block, and is sprung, thus providing a means for tensioning the detector element. Normally the dead space surrounding this spring is arranged to be on the downstream side of the gas flow, so that it does not contribute to the effective volume of the cell, but if in any particular application this volume is significant, it can be reduced to negligible proportions by being filled with packing except for the minute volume required for actual flexure of the spring. As an alternative means of supporting the resistance wire in each cell the gasket is fabricated of an insulating heat resistant material e.g. mica and the resistance wire of the detector element is 65 * Regd. Trade Mark.

attached to the gasket so that the wire traverses the perforation defining the cell in the final assembled apparatus, the gasket with attached wires thus forming an independent replaceable unit.

The resistance wire or wires may be fixed to the gasket in any suitable manner e.g. with a suitable adhesive. Furthermore the resistance wires can be etched from a single thin sheet of resistance material, e.g. nickel or platinum. In this case a spring loop can

be included in each etched wire.

In practice it is found necessary to arrange for the resistance wire to be stretched substantially across the centre of the volume space of the cell. This may be realised by fixing the wires in grooves in the gasket of depth equal to half its thickness. Preferably the wires may be fitted across one face of the gasket and in the final assembly a second gasket of the same thickness is sandwiched against the first so that the resistance wire stretches across the cell from between the two gaskets. This method automatically centres the wire in the cell.

Connection of the ends of the resistance wire through the cell block to means for measuring its electrical resistance e.g. a Wheatstone Bridge circuit may be effected by contact wires passing through insulators in one of the blocks. These may be constructed to protrude slightly above the surrounding face of the block through which they pass and are positioned so that on assembly of the apparatus by sandwiching of the gasket 100 bearing the resistance wires between the two cell blocks these wires are brought into physical contact with the ends of the resistance wires thereby completing the electric circuit with for example the Wheatstone 105 Bridge. Alternatively the ends of the wires passing through the insulators may be flattened and bent so as to press against the end of the resistance wire and in the case where two gaskets are employed to press 110 against the end of the resistance wire and lie between the two gaskets. In addition to this alternative the contact between the ends of the contact wires and the resistance wire may be made by the use of thin metal shims 115 similarly compressed between the gaskets.

The main advantages accruing from the method of construction in accordance with the invention may be summarised

follows: 1. Very small cell volumes are attainable (down to the order of 1 microlitre or less). and this, combined with the low thermal inertia of the fine wires used, allows the complete resolution of the closest fractions 125 emerging from a capillary gas chromatography column.

2. A further consequence of the very small volume is that the detector elements are close enough to the surrounding cell 130

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material to run relatively cool at high bridge currents. Thus a larger signal output per unit change of thermal conductivity can be obtained than with most other cells, and the chance of pyrolytic decomposition of samples is reduced.

3. The use of single wire detector elements, keeps noise due to movement of the filament to a very low level. The straight wire, rather than the more common coiled filament, also contributes to the cool running mentioned above.

4. A positive gas seal is obtained, enabling the cell to be used at high and low pressures without special precautions.

The apparatus of the present invention is used to determine the thermal conductivity of gases in combination with means for measuring the electrical resistance of the detector elements. Such means may be for example a Wheatstone Bridge circuit.

The apparatus of the present invention is described in more detail with reference to the accompanying drawings in which:

Figure 1 represents a side elevation in section through one of the cells of the 25 apparatus showing the method of construc-

Figure 2 represents a plan view in section about the line A—A¹ in figure 1

Figure 3 represents a side elevation in section through the same cell as Figure 1 having a different gasket and resistance wire supporting means

Figure 4 represents a plan view of a gasket having attached resistance wires and

Figure 5 represents schematically the apparatus of the invention in association with a Wheatstone Bridge circuit for measur-40 ing the thermal conductivity of a gas stream.

Referring to Figure 1 a particular embodiment of the apparatus of the present invention comprises Duralumin* blocks 1 and 2 separated by a perforated aluminium gasket 3. The gasket is sandwiched between two subsidiary aluminium foil gaskets 3a and 3b. As shown in figure 2 the perforations take the form of slots 4 which define narrow elongated cells 5 when combined with the duralumin blocks in the finished assembly.

The gasket is held between the blocks by screws 6 which pass through holes drilled in the assembly. The block 1 has drilled ports 55 7 providing for admission and withdrawal of gases from the cells. Each cell contains two platinum wires 8 running parallel the length of the cell. Thus space is saved by placing the elements forming two opposite arms of the Wheatstone Bridge within the volume of one cell, instead of in two separate cells in series as is normally done. Thicker support wires 8a pass through holes drilled in the other block 2 and are soldered to the

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ends of the platinum wires. These support wires are insulated from the block by a suitable insulating packing 10 e.g. a ceramic material. As shown in figure 2 the holes drilled in block 2 for the support wire insulation are staggered to allow the two wires 8 to run parallel, the cell being otherwise too narrow to allow room for this arrangement. In order to keep the detector wires taut within the cells, flexing of one of the support wires is allowed in the region 11 by omitting the insulating packing material from the top of one of the pair of holes drilled through block 2 for each support wire.

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Referring to Figure 3 this shows an alternative construction for the gasket and resistance wire supports, the gasket being formed as a laminar structure of two thinner gaskets 3c and 3d. In this case the ports 7 are chamfered at their inner edge as shown so as to direct the flow of gases through the cell. This arrangement also protects the resistance wires from any tubes etc. that might be inserted in the ports. The platinum wires 8 running parallel the length of the cell are cemented at each end to the gasket 3c. Thicker contact wires 8a pass through holes 9 drilled in the block 2 as before. These wires protrude above the surrounding face of the block and contact the ends of wire 8 by passing through holes 12 through the lower gasket 3d their ends being cylindrically curved to prevent damage to the resistance wires. In this embodiment the resistance wires can be tensioned by fine spring wires 100 or supported by quartz fibres (not shown) similarly sandwiched between the mica gaskets at the sides of the slots. A plan view of a gasket having resistance wires affixed is shown in Figure 4.

Referring to Figure 5, in operation the wires in each cell are connected as opposing arms of the Wheatstone Bridge which is itself connected to a source of supply of electric current e.g. an accumulator or DC 110 power supply 13. A potentiometer recorder 14 replaces the conventional galvanometer in the circuit.

The wires are heated by passage of electric current and the sample gas to be 115 analysed is passed through one cell and a suitable reference gas to balance out temperature changes is passed through the other cell. In chromatography work the reference gas usually takes the form of the carrier gas 120 employed e.g. nitrogen or helium. The degree of unbalance of the bridge represents a measure of the thermal conductivity of the sample gas. By use of the potentiometer recorder 14 the degree of unbalance of the 125 system can be shown visibly as a tracing on a moving chart. By continuously flowing the gas stream leaving a chromatographic column through one of the cells of the detector and a reference gas through the other and 130

employing the above system it is possible to obtain a tracing from the recorder showing peaks for every component of the gas mixture separated by the column. By adjusting the potentiometer recorder to show zero reflection with reference gas passing through both cells, under normal operation of the instrument the area under a peak shown on the tracing corresponds to the amount of a particular component in the gas mixture. This provides a ready method of repetitive analysis of the gas stream. It also provides a method for indicating when the gas stream passing through the detector cell contains the maximum proportion of a particular component. This information can be of use if it is desired to separate this component for further analysis. Other uses of the apparatus will be apparent to those skilled in the art. WHAT WE CLAIM IS:

1. Apparatus for use in the determina-tion of the thermal conductivity of gases which comprises two blocks of a thermally conductive material separated by a perforated gasket, the assembly of blocks and gasket defining one or more enclosed cells each cell having inlet and outlet ports with substantially the total gas space of each cell lying within the thickness of the gasket and at least one electrically heated detector element in each cell.

2. Apparatus as claimed in claim 1 wherein the gasket is formed as a laminar structure comprising a main body sandwiched

between two foils.

3. Apparatus as claimed in either claim 1 or 2 wherein a perforation in the gasket defines the size and shape of an enclosed cell.

4. Apparatus as claimed in any of the preceding claims wherein the detector element is a resistance wire stretched through the centre of the volume space of the cell.

5. Apparatus as claimed in claim 4 wherein the resistance wire is fixed at each end to electrically conductive support wires held normal to the resistance wire and passing through one of the blocks of thermally conductive material.

6. Apparatus as claimed in claim 5 wherein one of the support wires is free for part of its length within the block, and is sprung to provide means for tensioning the detector element.

7. Apparatus as claimed in claim 1 wherein the detector element is a resistance wire stretched through the centre of the volume space of the cell and is attached at

both ends to the gasket which is fabricated of an insulating, heat resisting material.

8. Apparatus as claimed in claim 7 wherein the ends of the resistance wire are held in physical contact within the assembled apparatus with electrically conductive wires held normal to the resistance wire and passing through one of the thermally conductive blocks.

9. Apparatus as claimed in claim 7 wherein the ends of the resistance wire are fixed in grooves of depth half the thickness

of the gasket.

10. Apparatus as claimed in claim 7 wherein the gasket is formed as a laminar structure comprising two thinner gaskets and the resistance wire is stretched across the cell from between the thinner gaskets.

11. Apparatus as claimed in claim 10 wherein the ends of the resistance wire are held in physical contact within the assembled apparatus with electrically conductive wires passing through one of the thermally conductive blocks and through the adjacent

thinner gasket. 12. Apparatus as claimed in claim 11 wherein the ends of the electrically conductive wires in contact with the ends of the resistance wire are flattened, the flattened ends lying between the two thinner gaskets within the assembled apparatus.

13. Apparatus as claimed in any one of the preceding claims 7 to 11 wherein the resistance wire is tensioned by spring wires

attached to the gasket.

14. Apparatus as claimed in any one of the preceding claims 7 to 12 wherein the resistance wire is etched from a single sheet of resistance material.

15. Apparatus as claimed in claim 14 wherein the etched wire includes a spring loop for tensioning the wire in the assembled apparatus.

16. Apparatus as claimed in any one of the preceding claims having two cells each cell containing two detector elements.

17. Apparatus for use in the determination of the thermal conductivity of gases 105 substantially as hereinbefore described and shown in the accompanying drawings.

18. A method of determining the thermal conductivity of gases employing apparatus as claimed in any of the preceding claims.

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Abingdon: Printed for Her Majesty's Stationery Office. by Burgess & Son (Abingdon), Ltd.—1966.
Published at The Patent Office, 25 Southampton Buildings, London, W.C.2,
from which copies may be obtained.

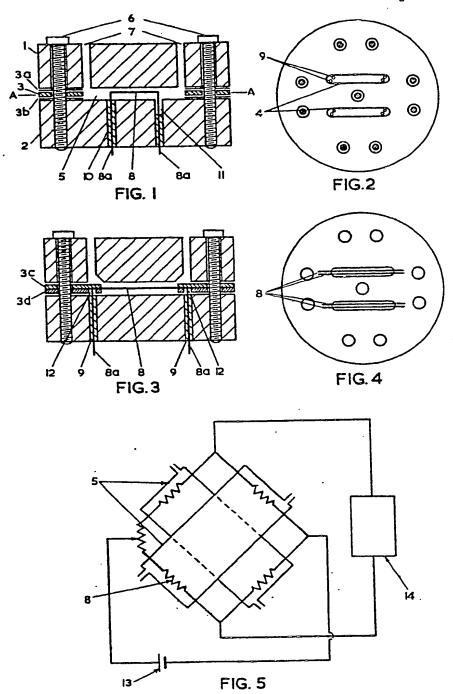
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1 SHEET

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